



## **Biosynthesis of Copper Nanoparticles Using Aqueous *Ficus Racemosa* Extract- Characterization and Study of Antimicrobial Effects**

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### **ABSTRACT**

Development of green nanotechnology is generating interest of researchers toward ecofriendly biosynthesis of nanoparticles. In this study, biosynthesis of stable copper nanoparticles were done using *Ficus Racemosa* leaf extract. First we prepared leaf extract of *Ficus Racemosa* sanctum in deionised water. This extract added to 0.02 M of copper sulfate solution and we observed the change in color of the solution from colorless to colored solution, this indicates that there is a formation of Cu nanoparticles. These biosynthesized Cu nanoparticles were characterized with the help of UV –Visible Spectroscopy. It was observed that the *Ficus Racemosa* leaf extract can reduce copper ions into copper nanoparticles within 8 to 10 min of reaction time. Thus, this method can be used for rapid and ecofriendly biosynthesis of stable copper nanoparticles. Further, microscopic examination and antimicrobial activity of synthesized nanoparticles was done against several bacteria and fungi.

**Keywords:** *Nano-biotechnology, Spectrophotometer, Nanoparticles, Physico-chemical properties, UV –Visible Spectroscopy, antimicrobial activity*

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## INTRODUCTION

Nanoparticles represent a particle with a nanometer size of 1–100 nm. The nanoscale material has new, unique, and superior physical and chemical properties compared to its bulk structure, due to an increase in the ratio of the surface area per volume of the material/particle [1].

A number of Indian traditional systems of medicines occur in India, of which prominent is Ayurveda. It's in existence for around 3000 years. Ayurvedic preparations have been successfully used for the treatment of various ailments from treating snake bites, to wound healing, with skin ointments protecting from infections, controlling anxiety, increasing memory and sharpening the overall beauty of a person.

The genus *Ficus* is an important group of trees which has various chemical constituents of promissive medicinal value. It is a sacred tree of Hindus and Buddhists. Four species of this genus constitute the group “Nalpamaram”, namely; *F. racemosa*, *F. microcarpa*, *F. benghalensis* and *F. religiosa* (Athi, Ithi, Peral and Arayal respectively).

*Ficus racemosa* is also known as *F. Glomerata*. *Ficus racemosa* has various synonyms like Udumbara (Udumbara is considered sacred to God Dattaguru), yajnanga, yajniya, yajnayoga, yajnyasara, gular, Cluster Fig tree, Country fig tree etc. It has been used in ritual sacrifice. It is one of the ksirivriksha – latex oozes out when the leaves are cut or plucked. It is one of the plants from a group, called pancavalkala, meaning the thick bark skins of five herbs, viz. udumbara, vata, asvattha, parisa and plaksa. The decoction of pancavalkala is used internally or for giving enema in bleeding per rectum and vagina (Raja Nighantu). Maharishi Charka has categorized udumbara as mutrasangrahaniya – anti-diuretic herb. Susruta has described the properties of the plant, like astringent, promotes callus healing in fractures (bhagnasandhaniya), alleviates Rakta pitta, burning sensation and obesity, and useful in vaginal disorder. Different parts of *Ficus Racemosa* are traditionally used as fodder, edible and ceremonial [2]. All parts of this plant (leaves, fruits, bark, latex, and sap of the root) are medicinally important in the traditional system of medicine in India. The astringent nature of the bark has been employed as a mouth wash in spongy gum and also internally in dysentery, menorrhagia and haemoptysis [3]. The bark is antiseptic, antipyretic and vermifugal, and the decoction of bark is used in the treatment of various skin diseases, ulcers and diabetes. It is also used as a poultice in inflammatory swellings/boils and regarded to be effective in the treatment of piles, dysentery, asthma, gonorrhoea, gleet, menorrhagia, leucorrhoea, haemoptysis and urinary diseases [4].



**Figure 1: Synthesis of Copper Nanoparticles**

## TAXONOMY

Kingdom	<i>Plantae</i>
Division	Magnoliophyta
Class	Magnolipsida
Order	Urticales
Family	Moraceae
Genus	<i>Ficus</i>
Species	<i>racemosa</i>

Various part of *Ficus racemosa* Linn are use in the treatment of various disease [5].

**Roots:** Roots are used in diabetes and pectoral complaints. Roots are also applied in mumps and other inflammatory glandular enlargements, also in hydrophobia

**Bark :** Bark is useful in diabetes, urological disorder and leprosy. It is highly effective in threatened abortion. The bark exhibited hypoglycemic effect in normal and alloxan-induced hyperglycemic animals and *b*- sitosterol-D-glucoside was identified as the active principle. The bark exhibited hypoglycemic effect in normal and alloxan-induced hyperglycemic animals and *b*- sitosterol-D-glucoside was identified as the active principle.

**Leaves:** The leaves are good for washing the wounds and ulcer. They are also useful in diarrhea and dysentery.

**Fruit:** The fruit are useful in the treatment of leucorrhoea and blood disorder. They are also useful in miscarriage, menorrhagia, cancer and visceral obstructions.

## Need For Green Synthesis

Green synthesis of nanoparticles is a bottom up approach where the reaction occurring is reduction/oxidation. Nanoparticles can be prepared by different methods. Chemical approaches

are the most popular methods for the production of nanoparticles. However, some chemical methods cannot avoid the use of toxic chemicals in the synthesis protocol. The chemically synthesized nanoparticles carry some reactive functional groups, which can be toxic to biological system. Hence, the development of clean, biocompatible, non-toxic and eco-friendly methods for the synthesis of nanoparticles deserves merit. The green synthesis of metallic nanoparticles (NPs) has attracted tremendous attention in recent years because these protocols are low cost and more environmentally friendly than standard methods of synthesis [6].

## MATERIALS AND METHOD

All the chemical reagents used in this experiment were of analytical grade purchased from Himedia. The *Ficus racemosa* leaves were collected from in and around Lucknow, Uttar Pradesh. Thoroughly washed leaves (5g) were cut and boiled with 20 ml of de-ionized water for 15 min in heating mental at temperature 80°C. The resulting product was filtered and stored in refrigerator for further experiments.

### Preparation of Plant Leaf Extract

The freshly chopped *Ficus racemosa* leaves were weighed 5g and mixed in 20ml of distilled water and boiled for 15 min at 80° C. The extract was allowed to cool at room temperature and filtered by Whatmann No1filter Paper. The filtrate was stored in a beaker tightly seal packed with aluminium foil and paraffin and kept at room temperature for further use. 20ml of extract was placed in a clean and dry beaker and heated to dryness over water bath [7].

### Phytochemical Tests [8]

**Alkaloid Test:** Wagner's reagent was prepared by adding 2g of iodine and 6g of KI in 1000ml of water and then 1 ml of plant extract was added to 1ml of above reagent. A reddish brown precipitate indicates the presence of alkaloids in the plant extract.

**Saponin Test:** 1ml of plant extract was added to 5 to 10ml of distilled water. The test tube was shaken well to note a stable froth. The froth formation indicates the presence of saponin in the plant extract.

**Tannin Test:** 5% ferric chloride solution was prepared. 2ml of 5% ferric chloride solution was added to 1ml of plant extract. The color change which gives blue, black or dark green color indicates the presence of tannin in the plant extract.

**Flavonoid Test:** 1% NaOH was prepared by dissolving 10g of sodium hydroxide in 1L of distilled water. 2ml of 1% NaOH was added to 1ml of plant extract. The yellow coloration indicates the presence of flavonoid in the plant extract.

**Starch Test:** 2 drops of iodine solution was added to 2 ml of plant extract. The blue-black colour observed indicates the presence of starch in the plant extract.

**Carbohydrate Test:** Fehling A was prepared by adding 7g of copper sulphate in 100ml distilled water with 2 drops of sulphuric acid to the reagent. Fehling B was prepared by adding 35g of potassium sodium tartarate and 12g of sodium hydroxide in 100ml of distilled water. 0.5ml of Fehling A and Fehling B was added to the 0.5ml of plant extract. The mixtures were then boiled and the formation of brick red precipitate of cuprous oxide indicates the presence of carbohydrate in the plant extract.

**Protein Test:** 0.5% of copper sulphate and 0.5% of 10% NaOH was added to 1ml plant extract. The purple coloration indicates the presence of protein in the plant extract.

**Fat Test:** 1ml of distilled water and few drops of ethanol were added to 1ml of plant extract. The white colour precipitate indicates the presence of fat in the plant extract.

### **Preparation of 0.02M Solution of Copper Sulphate**

The solution used in synthesizing copper nanoparticles was analytical grade copper sulphate. A 0.02M stock solution of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  in chloride free distilled water was prepared. 0.031g of Copper Sulphate was transferred in a 50-ml beaker and 20ml de-ionized water was added dropwise while swirling to dissolve the salt. The beaker containing solution was tightly covered with aluminium foil and paraffin and kept away from light for further use[9].

### **Synthesis of Copper Nanoparticles:**

0.02 M copper sulphate was prepared in 5ml distilled water for each synthesis and blue solution was seen. Glycerine (500 $\mu\text{l}$ ) was added to aqueous solution containing copper sulphate with vigorous stirring. 10ml plant extract was added to copper sulphate solution with glycerine. Sodium hydroxide (500 $\mu\text{l}$ ) was added dropwise under rapid stirring. Colour changes from light to dark green. Reduction started and greenish brown colour formation confirms formation of copper nanoparticles.

### **Purification of Nanoparticles**

The reaction mixture was centrifuged at the speed of 5000 rpm and the supernatants were taken out by aspiration. This clear light green liquid was analysed by a UV-Visible spectrophotometer. The content/residue was washed in double distilled water thrice, centrifugation was repeated at 5000 rpm for 30 minute and powder was collected and dried in hot air oven at 110°C for 3 hours. These oven-dried particles were used for further analysis for determining size, shape and chemical composition.

### **Characterization of Nanoparticles:**

UV - Visible spectrophotometer: The peak of the nanoparticles should be in the range of 400-600 nm. Best nanoparticles are synthesized at 420 nm having sharp peak. Microscopic examination of nanoparticles: was done by mounting the nanoparticles with water and safranin and visualizing under the light microscope[10].

### Antimicrobial Activity

#### Test organism for antibacterial activity

Gram positive (+) bacteria and gram negative (-) bacteria, and fungus were used for determining antimicrobial activity. The bacterial pathogens were collected from the Department of Biotechnology, MNNIT. The bacteria used are as follows

#### List of microbes:

S No.	Name of the Microbes	Bacterial type
1	<i>E. Coli</i>	Gram negative
2	<i>S.aureus</i>	Gram positive
3	<i>B.subtilis</i>	Gram positive
4	<i>A.niger</i>	-
5	<i>F.Oxysporum</i>	-

The bacterial and fungus strains were grown and maintained on nutrient agar and potato dextrose agar slant at 37°C, followed by incubation for 5 days. The culture was stored at 4°C for further use. The organisms were sub cultured once in every 15 days [11].

#### Media preparation

In this study, Nutrient Agar and Potato Dextrose Agar medium was used which supports growth of a wide range of bacteria including *Bacillus subtilis* and *Escherichia coli*. For preparing nutrient agar and Potato Dextrose Agar medium both the powders were accurately weighed and were dissolved in distilled water. The medium was kept in cotton-plugged glass container, sterilized in an autoclave at 121°C for 15 mins. It was distributed inside a laminar hood on petri dishes hot 45° C and allowed to solidify.

#### Method for Testing Antimicrobial Activity of Synthesized Copper Nanoparticles

Antimicrobial activity of biological synthesized Copper nanoparticles was carried out by Agar well diffusion method against Gram negative and Gram positive bacteria and fungus. The pure culture of organism was sub cultured in nutrient broth and potato dextrose broth. The nutrient agar plates were prepared by 25ml (for one plate) of molten media into sterile petri-plates. For bacterial and fungal growth, a culture was prepared by spreading the 30µL fresh culture with each test organism on nutrient agar plates with the help of a sterile glass rod spreader. Plates

were left standing for 10 minutes to let the culture get absorbed. Then 6 mm (size) wells (4 well) were punched at four corners of each Petri dish at 1.5 cm away from the wall into nutrient agar plates for testing antimicrobial activity. Using the micro-pipette 50µl, 75µl and 100µl of sample of nanoparticle suspension was poured into each well on the plates. Then antibiotic-Ciprofloxacin was used as positive control against bacteria and antifungal drug fluconazole was used as positive control against different fungus. Using the micropipette, 10µl of positive control solution was poured into one of the well. After adding the samples in the wells, the dishes were kept in a refrigerator for an hour for absorption of the samples into the surrounding medium from the well. The plates were transferred into an incubator set at 37°C to allow bacterial growth on the medium. After 24 hrs the plates were taken out of the incubator and observed for zone of inhibition around the wells. The zone of inhibition was measured in millimetres [12].

## RESULTS AND DISCUSSION:

### Biosynthesis of copper nanoparticles from ficus plant extracts.

In this study, copper nanoparticles were synthesized using ficus leaf extract under static condition. It is well known that copper nanoparticles exhibit dark green colour in aqueous solution due to excitation of surface plasmon vibrations in copper nanoparticles. As the extract was mixed in the aqueous solution of the copper ion complex, it started to change the colour from colourless to greenish brown due to reduction of copper ion which indicated formation of copper nanoparticles (Fig. 1).

### Phytochemical analysis of ficus leaf extracts

The phytochemical characteristics of the medicinal plants investigated that are summarized in Table 1.

**Table 1: Qualitative analysis of phytochemicals in the extract of *ficus racemosa***

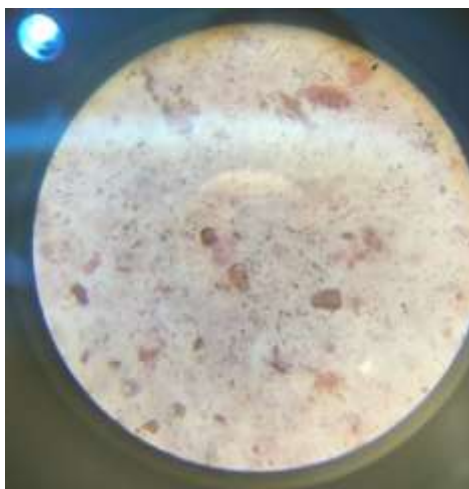
Phytochemicals	Test	Observation	Inference Presence (+), Absence(-)
Alkaloid	Wagner's test	Red precipitate	+
Tannin and Phenolic compound	Lead acetate test	Green colour	+
Terpenoid and Phytosterol	Salkowaski test	Reddish-brown colour	+
Saponin	Foam test	Presence of emulsion	-
Flavonoid	Ferric chloride test	White precipitate	-
Glycoside	-	Brown ring	-
Carbohydrates	Fehling's test	-	+
Test for starch	Iodine test	-	-
Test for proteins	Biuret's test	-	-
Fixed oil and fatty acid	Spot test	Presence of spot	+

The results revealed that the medicinally active constituents such as phenol, alkaloids, Terpenoids, carbohydrates and fats were present in ficus leaf extract. While saponin, flavonoid, glycoside, starch and proteins were absent in ficus leaf extract.

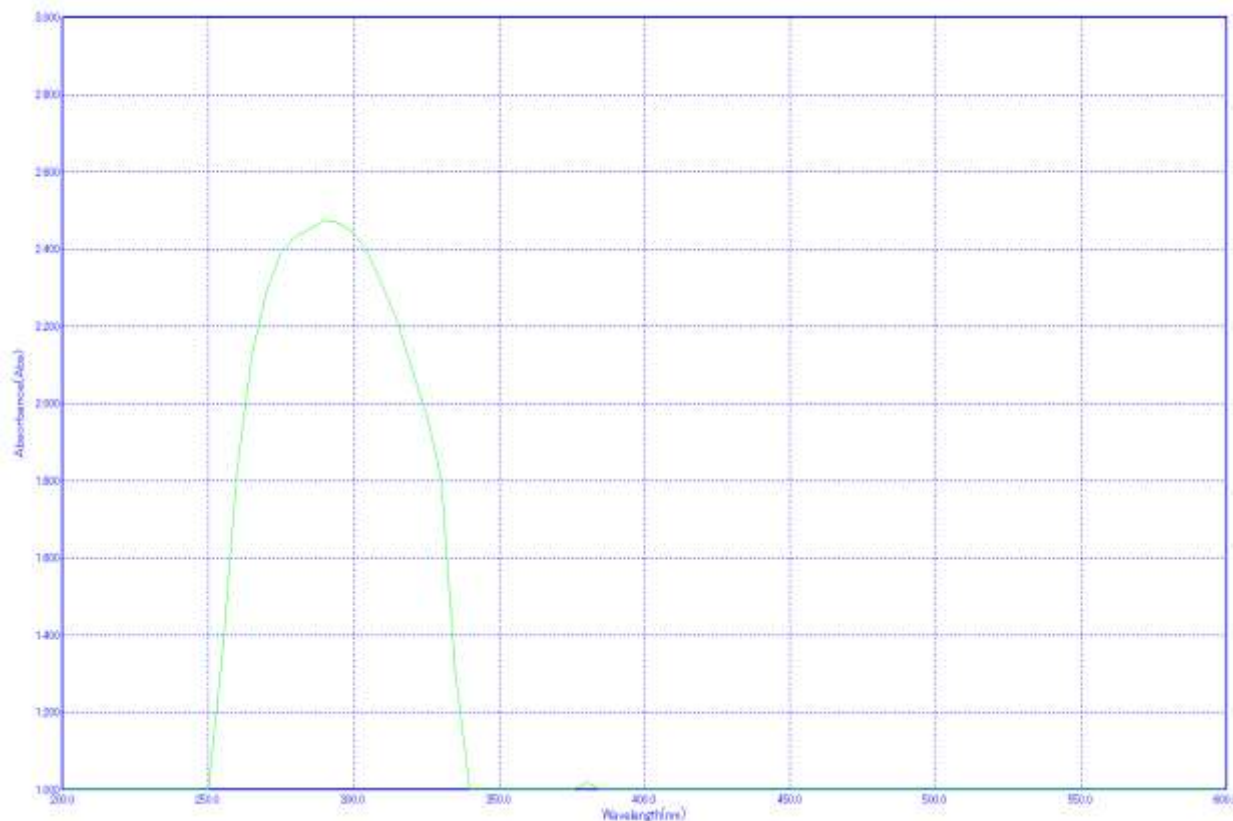
#### UV-Vis spectra analysis

The UV-Vis spectroscopy recorded from the nanoparticle solution showed the characteristic surface plasmon resonance (SPR) spectra with absorbance at 250 – 700 nm and peak maximum at 280 nm which is attributed to the formation of Cu nanoparticles. The formation of the copper nanoparticle was considered successful by initial change in colour. (Fig. 3).

Microscopic examination of nanoparticles: was done by mounting the nanoparticles with water and safranin and visualizing under the light microscope which showed the pink colour structure of cells. (Fig. 2).



**Figure 2: Microscopic examination of nanoparticles of *Ficus racemosa***

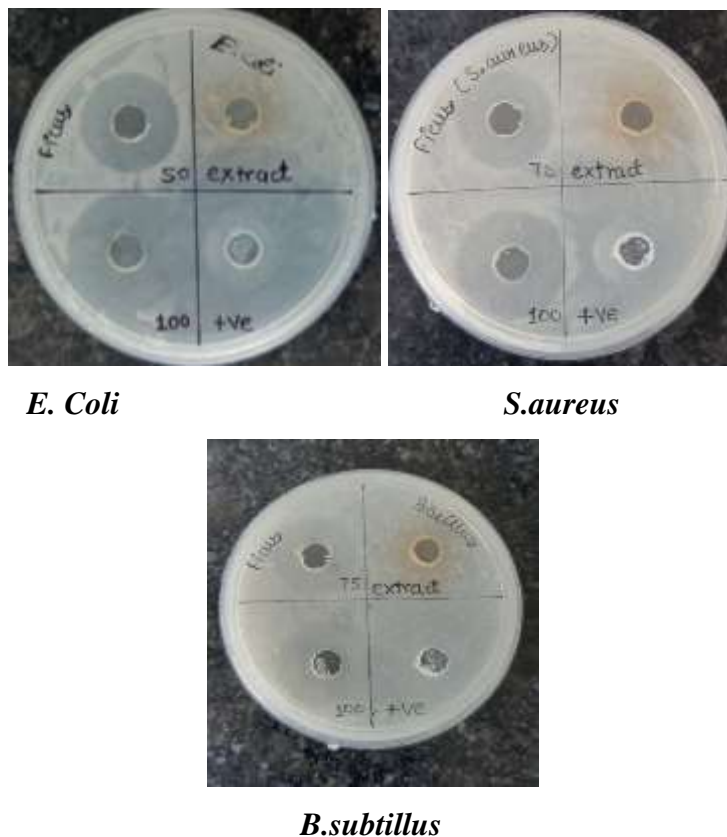


**Figure 3: UV-Visible spectroscopy result for *Ficus racemosa***

#### **Antibacterial activity:**

The antimicrobial activity of *Ficus racemosa* stabilized copper nanoparticles against three pathogenic bacteria *Bacillus subtilis*, *Escherichia coli* and *S.aureus* and two fungus *A.niger* and *F.Oxysporum* were evaluated and was compared to a commercial antibiotic Ciprofloxacin and antifungal drug fluconazole. The distinct Zone of Inhibition was observed around the well where in the suspension of CuNPs was applied. (Table 1).

The size of inhibition zone differed according to the type of microbes and the size and concentrations of copper nanoparticles. The zone of inhibition was observed to be more in gram positive bacteria when compared to gram negative bacteria. This is mainly due to the differences in pathogen's membrane structures. The anti-bacterial activity of copper nanoparticles showed more inhibition than that of plant extracts (control). For copper nanoparticles, zone of inhibition was found to be 10 mm, 10 mm and 9 mm against *Escherichia coli*, *Bacillus subtilis* and *Staphylococcus aureus* respectively. The highest zone of inhibition has reported by the synthesized copper nanoparticles against *Escherichia coli* and *Bacillus subtilis* with 100 $\mu$ l concentration of CuNPs . (figure 4).

*E. Coli**S.aureus**B.subtilis***Figure 4: Antibacterial activity of CuNPs from ficus leaf extract****Anti-fungal Activity of ficus leaf**

The results of anti-fungal activity were show in Table 2.

**Table 2: Antimicrobial activity of CuNPs from ficus leaf extract**

Concentration( $\mu$ l)	Diameter of zone of inhibition(mm) bacteria				
	<i>E. coli</i>	<i>B.subtilis</i>	<i>S.aureus</i>	<i>A.niger</i>	<i>F.Oxysporum</i>
50 or 75	8	8	7	12	3
100	10	10	9	13	6
Extract	7	7	5	7	0
Positive	16	20	10	10	7

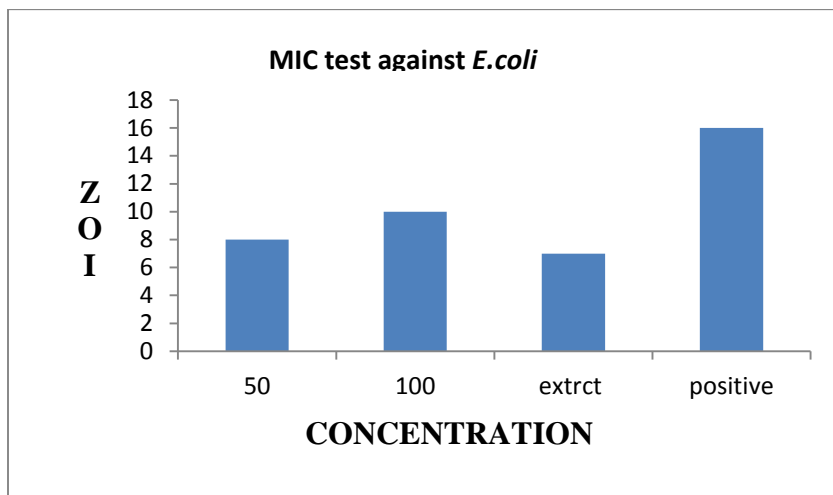
Zone of inhibition was observed with the ficus leaf extract, flucanazole as a control and copper nanoparticles against fungal strain (*Fusarium* species and *Aspergillus* species). The anti-fungal activity of copper nanoparticles showed more activity than that of plant extracts. For copper nanoparticles from ficus leaf extract the zone of inhibition was found to be 13 mm for *Aspergillus* species and 6 mm for *F.oxysporum* species. The highest zone of inhibition has reported by the synthesized copper nanoparticles against *Aspergillus* species. On the other hand, plant extract did not exhibit sufficient antifungal activity. (figure 5).



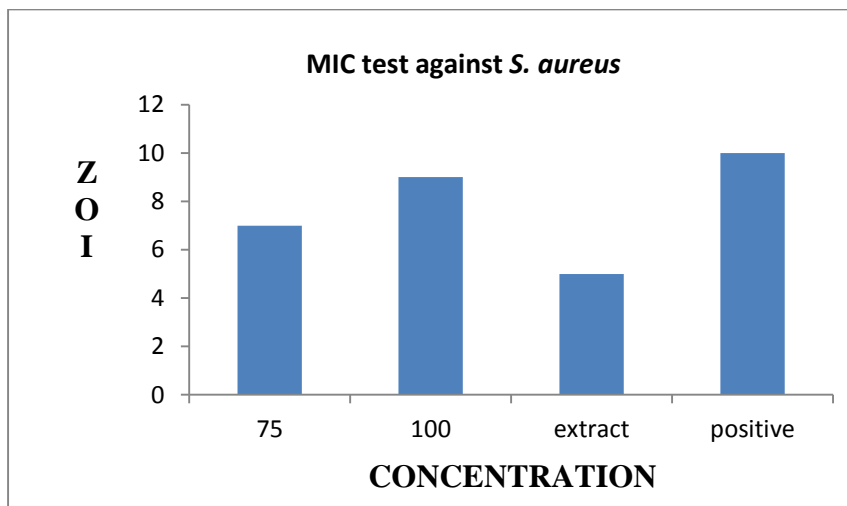
*A.niger*

*F.oxysporum*

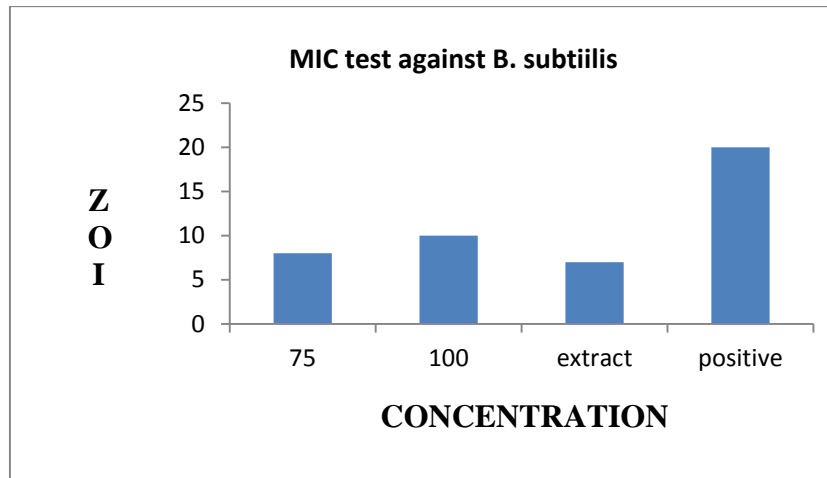
Figure 5: Antifungal activity of CuNPs from ficus leaf extract



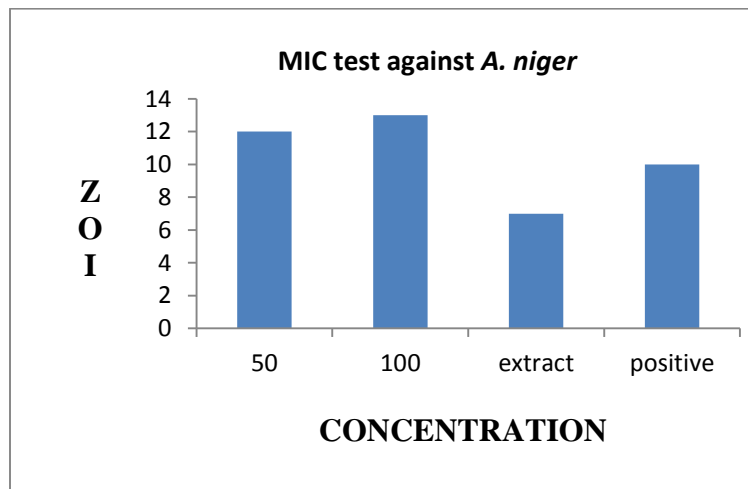
Graph 1: MIC against *E. Coli*



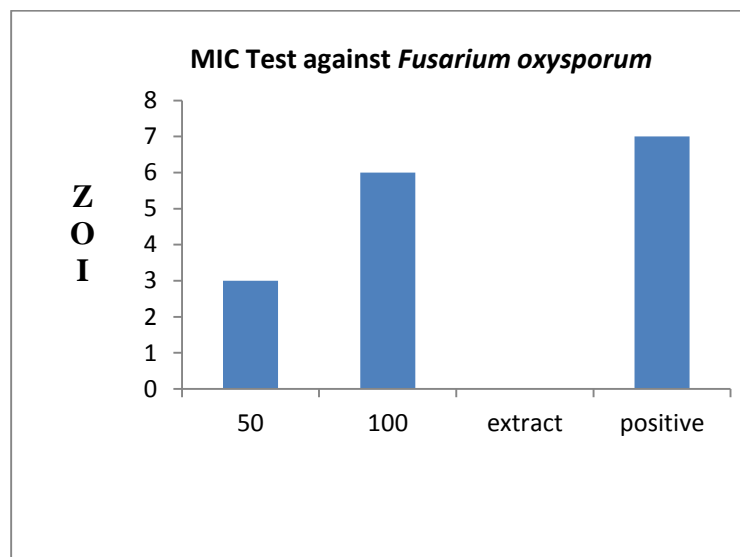
Graph 2: MIC against *E. Coli*



**Graph 3: MIC Test against *B.subtilis***



**Graph 4: MIC Test against *A.niger***



**Graph 5: MIC Test against *Fusarium Oxysporum***

## CONCLUSION:

Synthesis of nanomaterial with the desired quality and properties is one of the key issues in current nanotechnology. Today, the green synthesis of metallic nanoparticles has received increasing attention due to the development of eco-friendly technologies in materials science. Use of natural plant extracts in the preparation of nanoparticles by greener route provides advancement over chemical and physical method as it is cost effective, environment friendly. In conclusion, here we report green and biological synthesis of Cu nanoparticles using leaf extract of *Ficus racemosa*. This method has advantageous over other reported methods are easily available starting materials, inexpensive and process is simple to carry in any college level laboratory, use of toxic reagent is avoided and pollution free.

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